

# Passive Treatment Systems for the Removal of Selenium: Barrel Substrate Studies, Design, and Full-Scale Implementation



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**CH2MHILL®**

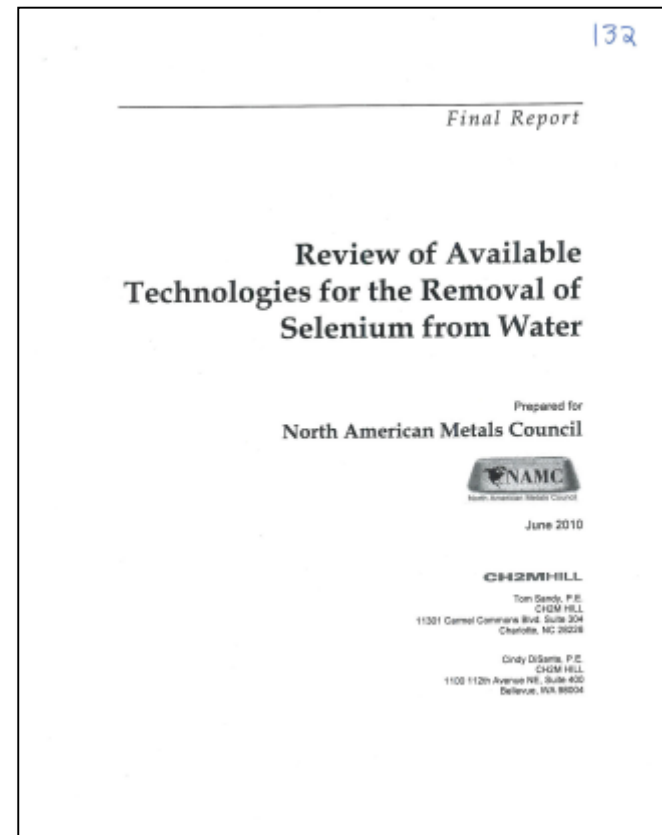
*June 3 2013*

# Overview of Presentation

## Outline

- Selenium Issues
- Mechanisms of Selenium Removal
- Historical Systems
- New Full Scale Systems
- Non-mining Examples
- By-Product Treatment
- Conclusions

## NAMC Selenium Report 2010



<http://www.namc.org/docs/00062756.PDF>



# Key Points about Selenium Control

- Se control can be critical to agriculture runoff, power generation, and mining industries
- Includes active and passive control technologies
- Passive treatment:
  - Uses natural processes to reduce and capture Se
  - Requires less operational effort and management
  - Demonstrated as a viable option for Se treatment



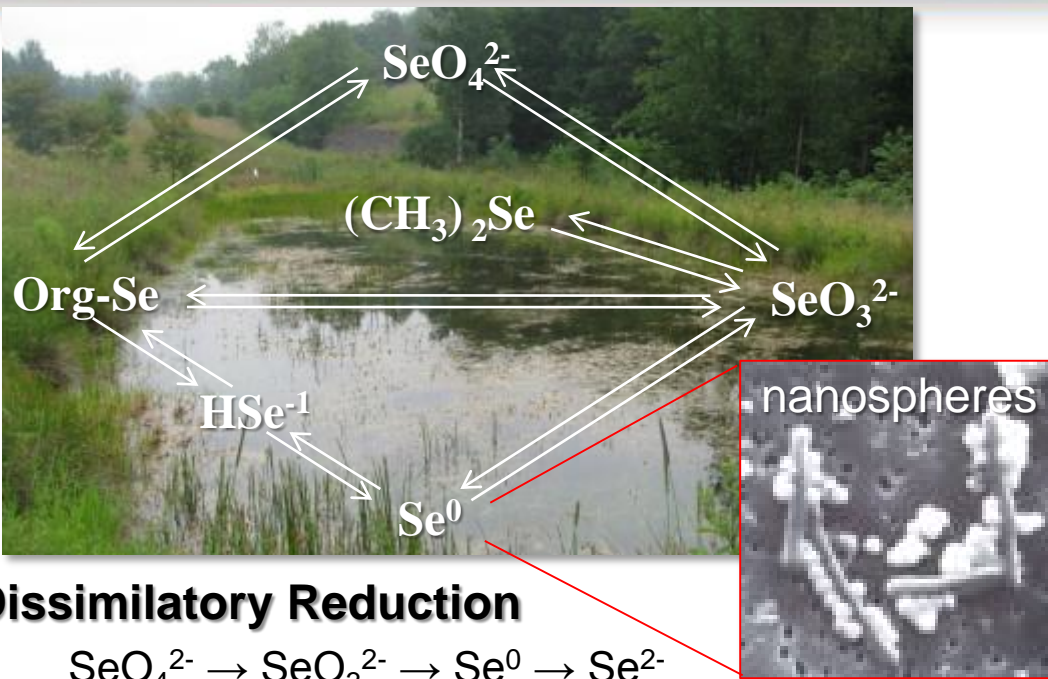
# Selenium Chemistry and Toxicity

- **Trace concentrations are essential for diet**
- **Large concentrations can be toxic**
  - Bioaccumulates; aquatic bird egg hatchability, fish larval deformities
- **Oxidation state determines bioavailability, toxicity**

– Elemental ( $\text{Se}^0$ )	Not toxic, unavailable
– Inorganic selenide ( $\text{Se}^{2-}$ )	Less bioavailable
– Selenate ( $\text{SeO}_4^{2-}$ , $\text{Se}^{6+}$ )	Soluble, bioavailable, less toxic
– Selenite ( $\text{SeO}_3^{2-}$ , $\text{Se}^{4+}$ )	Soluble, bioavailable, more toxic
– Organic selenide (org-Se)	Most bioavailable, most toxic
- **State NPDES effluent limits**
  - 4.7  $\mu\text{g/L}$  average month; 8.2  $\mu\text{g/L}$  daily max



# Wetland Processing and Storage of Selenium



## Volatilization

- ❑ Organic +  $\text{SeO}_3^{2-} \rightarrow (\text{CH}_3)_2\text{Se}$
- ❑ Volatilized from plant tissues
- ❑ 5-30% cumulative loss from sediments and plants

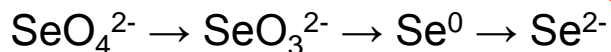
## Sorption

- ❑ Selenite sorbs to sediments and soil constituents:  $\text{Fe}^-$ ,  $\text{Mn}^-$  or  $\text{Al}$ -oxyhydroxides and organic matter

## Plant Uptake

- ❑ Rapid uptake
- ❑ Tissue concentrations increase but not detrimental
- ❑ No long term storage in plants; Se transferred to sediments

## Dissimilatory Reduction

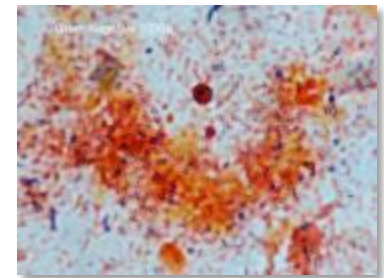
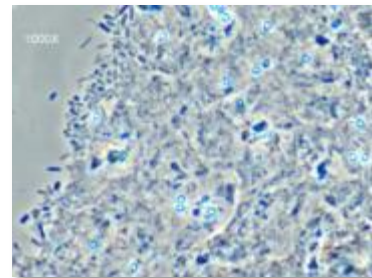
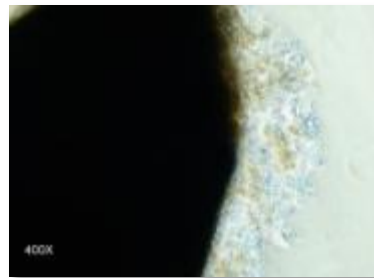
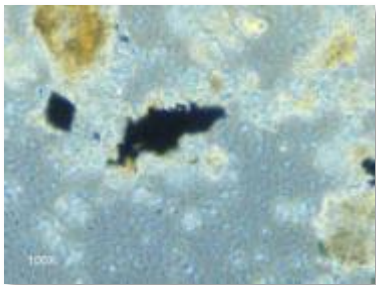


- ❑ Anaerobic process (Eh -200 mV, DO < 2)
- ❑ Distribution in wetland sediments:
  - ❑ 0:13:41:46
- ❑ *Wetlands*: 90% reduction 10 - 16 days
- ❑ *Bioreactors*: 90% reduction < 1 - 2 days

# Biological Selenium Treatment

■ Organics + Selenite/Selenate + N + P  $\longrightarrow$  New Cells + CO<sub>2</sub> + H<sub>2</sub>O + Se<sup>0</sup>

■ Order of reduction:



# Selenium Passive Treatment Systems: Free Water Surface Wetlands Provide Starting Point



Hansen et al, 1998

- Area: 36 ha
- Flow: ~6,540 m<sup>3</sup>/d
- Date: since 1991
- HRT: 7-10 days
- Se reduction: 89%
- Se in: 20-30 µg/L
- Se out: <5 µg/L
- Volatilization: 10-30%

# Early Passive Treatment Data

Site/ Date	GPM	pH	Influent Se ug/L	Effluent Se ug/L	Percent Removal
NV Gold Tailing (Aerobic) 1994	10	7.5	40	16	60%
NV Waste Rock (BCR) 1994	6	2.7	22	<5	>78%
Brewer Mine (BCR) 1995	1	2	1,500	50	97%

Source: Gusek (2013)  
CLU-IN Bioreactor Overview



# Great Recent Progress in Bioreactor Design



[www.sdcornblog.com](http://www.sdcornblog.com)

**Denitrifying Bioreactors for Agricultural Wastewater Treatment**



**Full-Scale Sequential Systems**

Bob Nairn



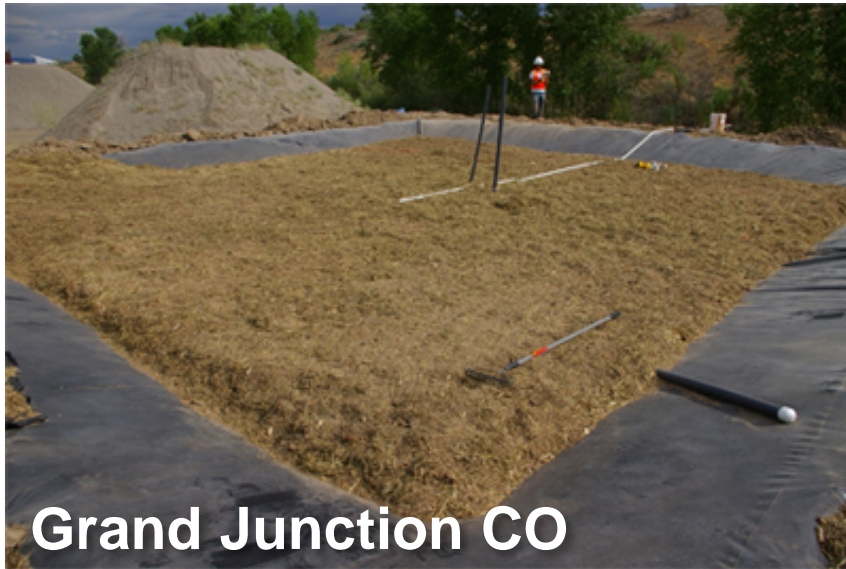
**Pilot Projects**

[www.nps.gov](http://www.nps.gov)

**Bioreactors for Mine Water Treatment**

# Case History

## Anaerobic “Bioreactor” Wetland Demonstration Showed High Efficiency in Minimal Area



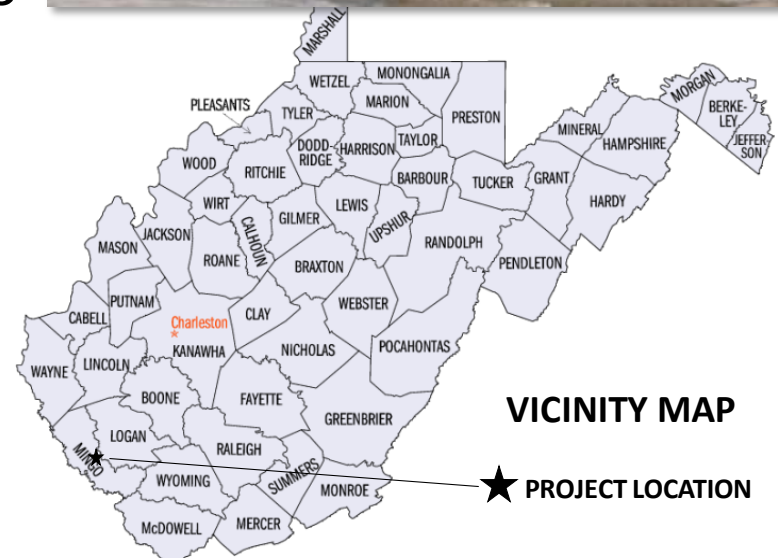
- Source: gravel pit seep
- Volume: 4,380 ft<sup>3</sup>
- Flow: 2-24 gpm
- Date: 9/08-10/09
- HRT: 2.4 d
- Se in flow: 1-34 µg/L
- Se reduction: 98%  
(90% winter)
- Se removal rate: 16 mg/d/m<sup>3</sup>
- Se out: 0.5 µg/L
- TCLP <1 µg/L Se

# Case Histories: Pilot and Full-Scale Passive Treatment in WV

## Overview

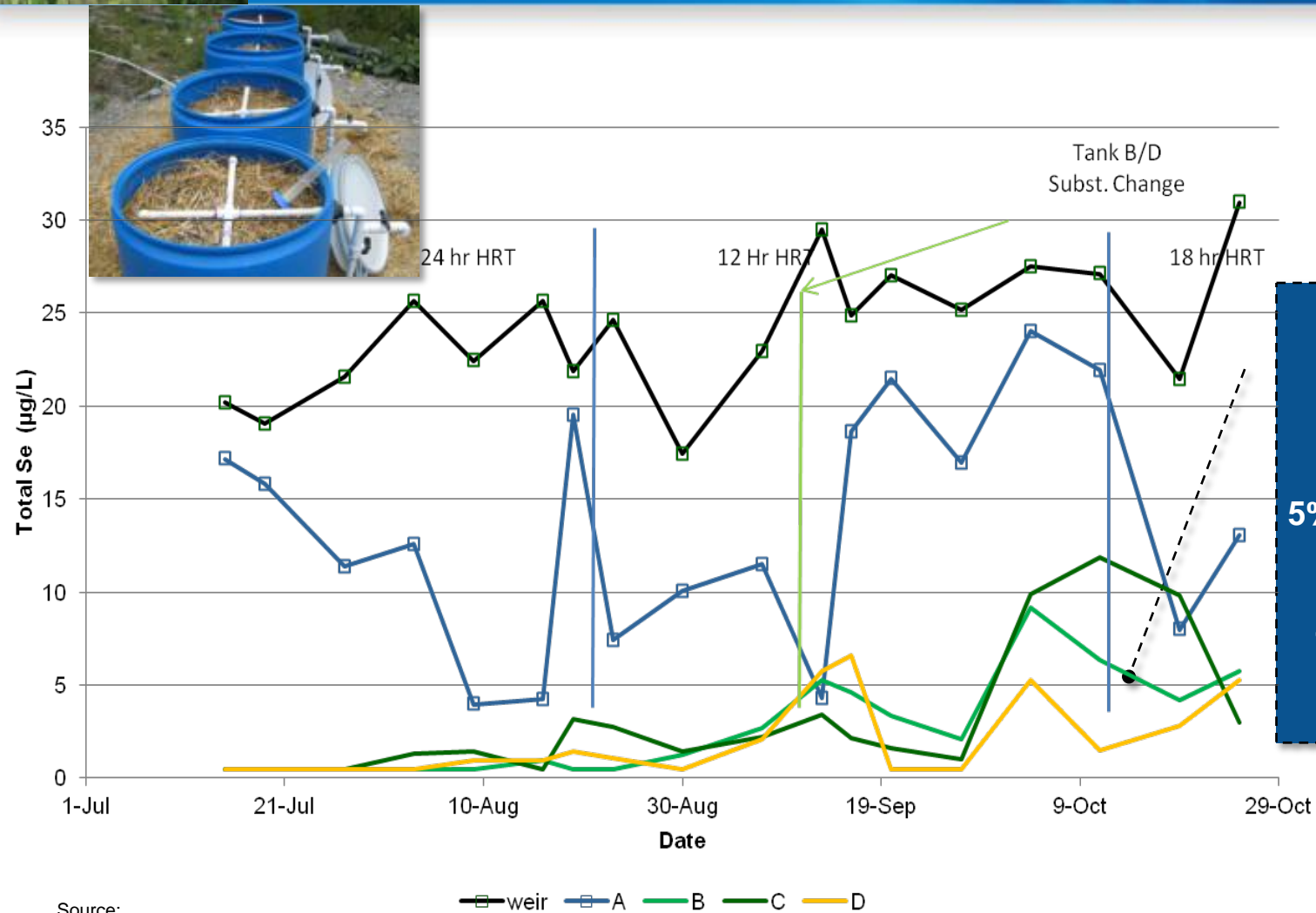
- Two outlets assigned stringent selenium discharge standard:
  - 4.7 ug/L monthly mean
  - 8.2 ug/L daily max
- Conducted barrel studies to formulate substrate, calibrate model
- Designed two distinct systems based on landscape, space, treatment
- First system July 2011
- Second system November 2011

## Location





# Case History (2010) Pilot Study of BCR for Coal Mine Drainage



**20% woodchips**  
**35% sawdust**  
**10% peat moss**  
**5% limestone sand**  
**25% hay,**  
**5% composted manure**

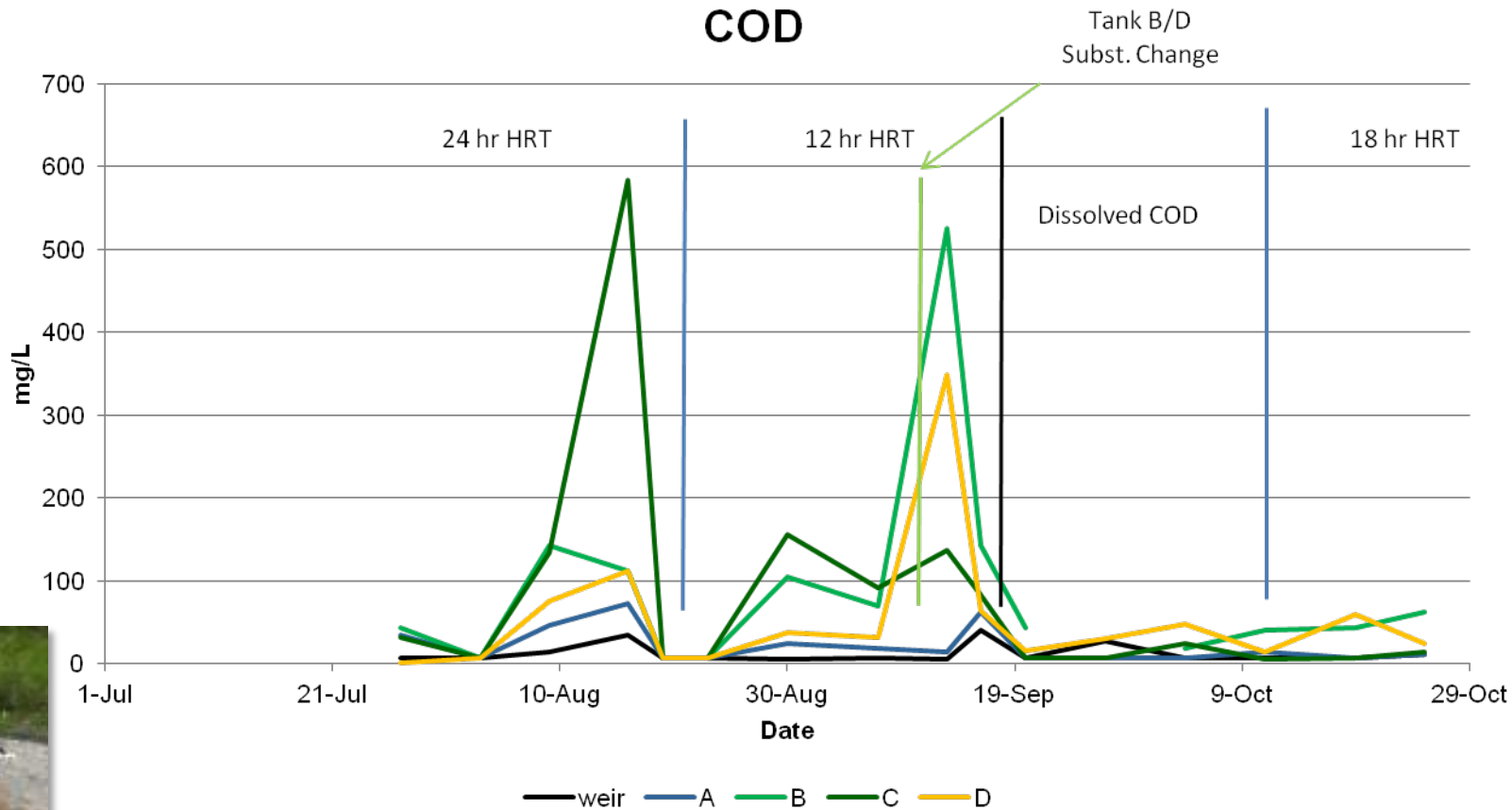
**24 hr HRT**  
**~ 17 mg/d/m<sup>3</sup> media**

**k<sub>20</sub> ~1700 m/yr**

Source: CH2MHILL (2010)



# Barrel Study Confirmed Significant Post-Startup "Byproduct" Discharges





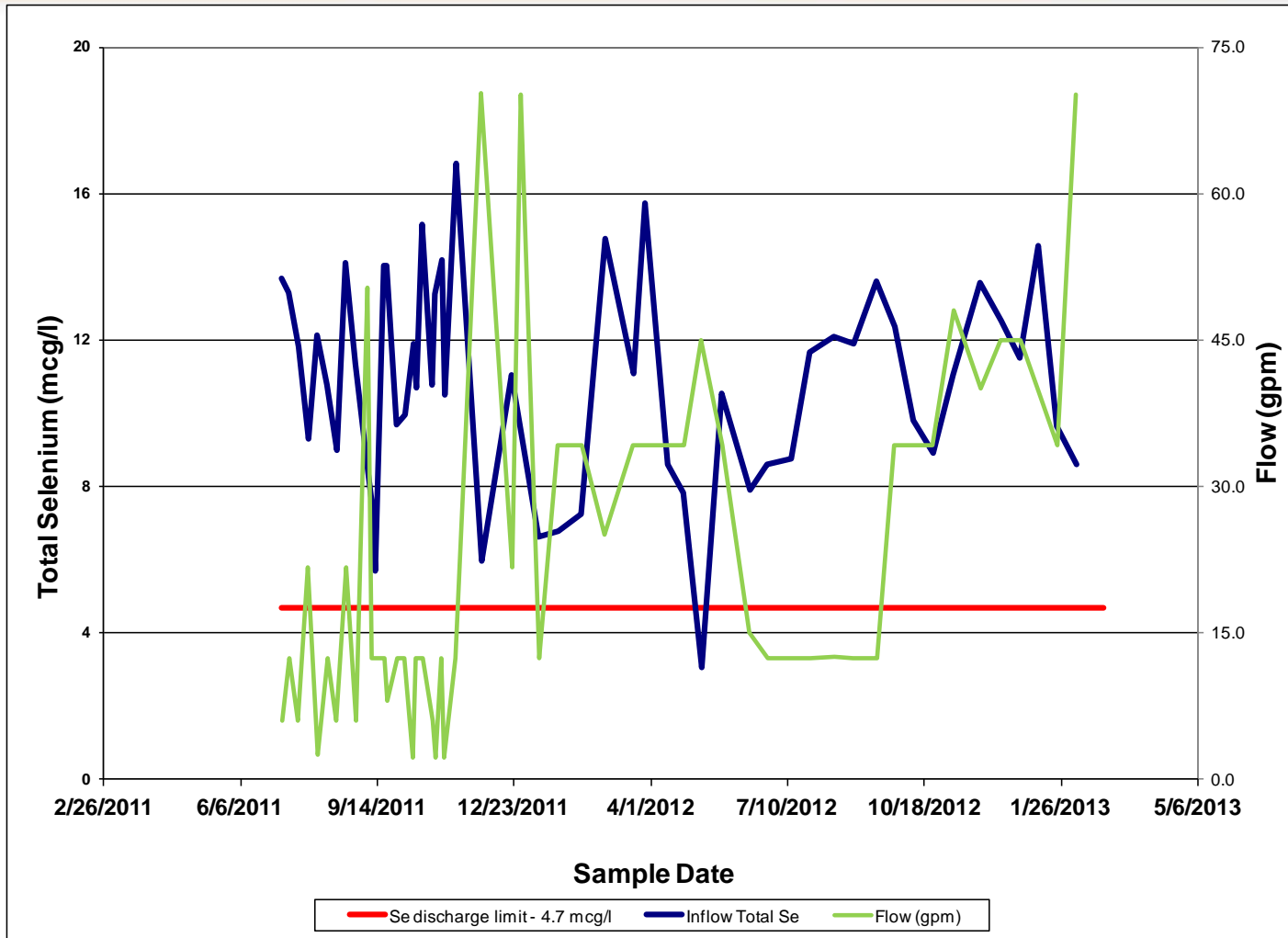
# Barrel Study Conclusions

- High strength substrate can remove Se at 12-hour HRT (24- to 48-hour HRT for typical design)
- High strength substrate initially generates elevated concentration of secondary parameters (BOD, COD, low DO, etc.)
- Low strength substrate = lower Se removal rates *but also* lower secondary parameters
- Initial Se removal is largely as weakly adsorbed selenite with minor amounts removed as elemental Se
- Recommend additional long-term investigation



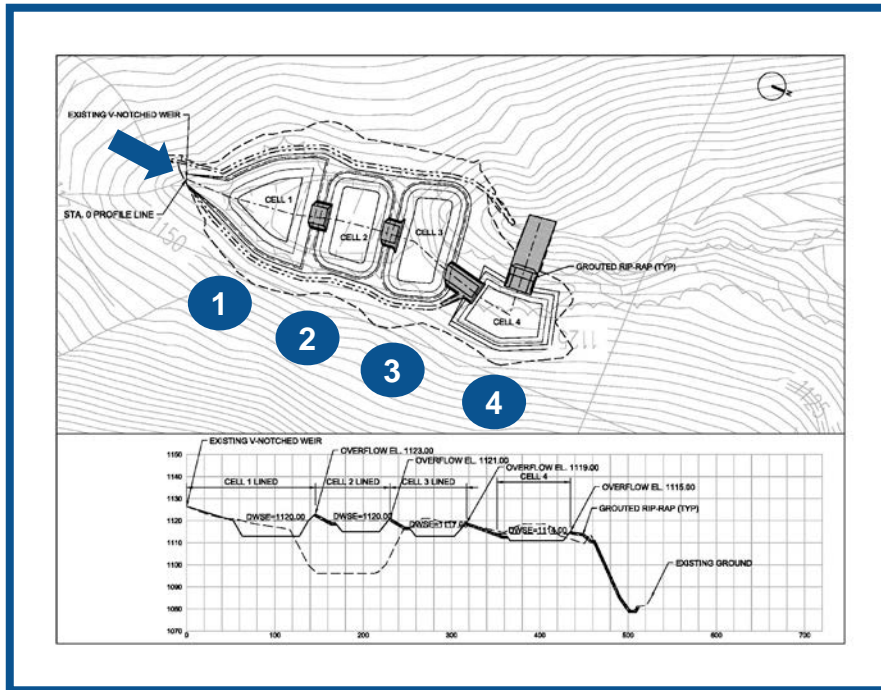
100 m

# System A: Design Flow Set to Capture Load and Account for Inter-annual Variation





# Case History (2011-present) Two Full-Scale BCR Systems for Coal Mine Drainage Treatment

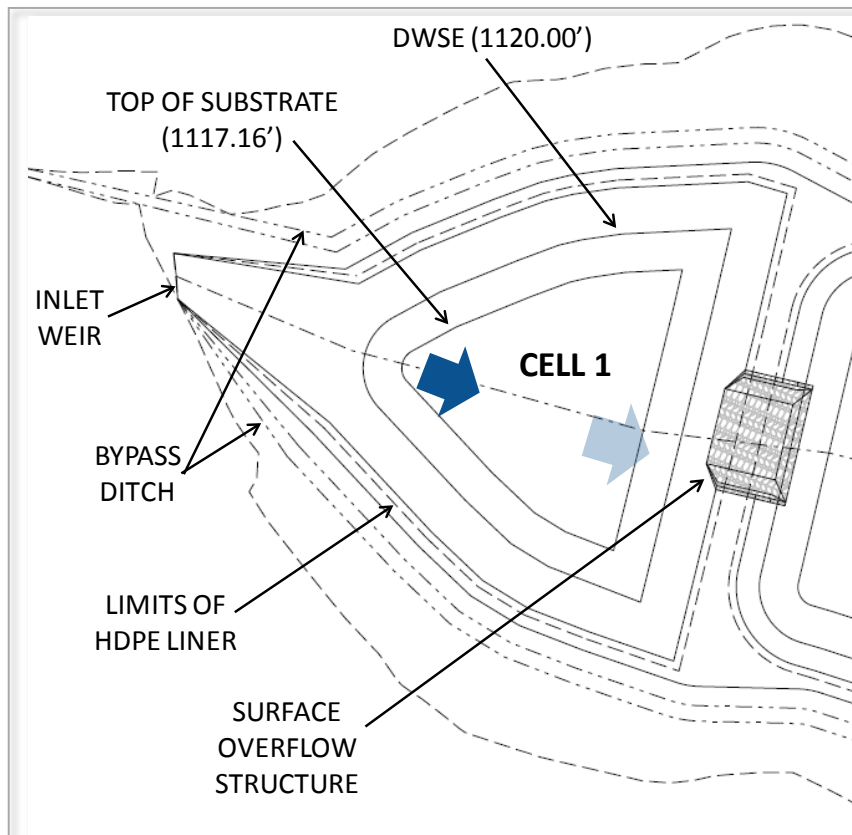


- 60 gpm base flow
- 100 gpm max
- 12  $\mu\text{g/L}$  mean Se to  $<4.7$

- Replace existing sed pond
- Four cells-in-series:
  1. 0.13 ac Downflow BCR  
Barrel "B" mix
  2. 0.14 ac Anaerobic upflow bed  
Barrel "A" peat
  3. 0.16 ac Fill-and-drain wetland  
Gravel; siphon level control
  4. 0.11 ac Surface flow marsh

# Cell 1: Downflow Biochemical Reactor (BCR)

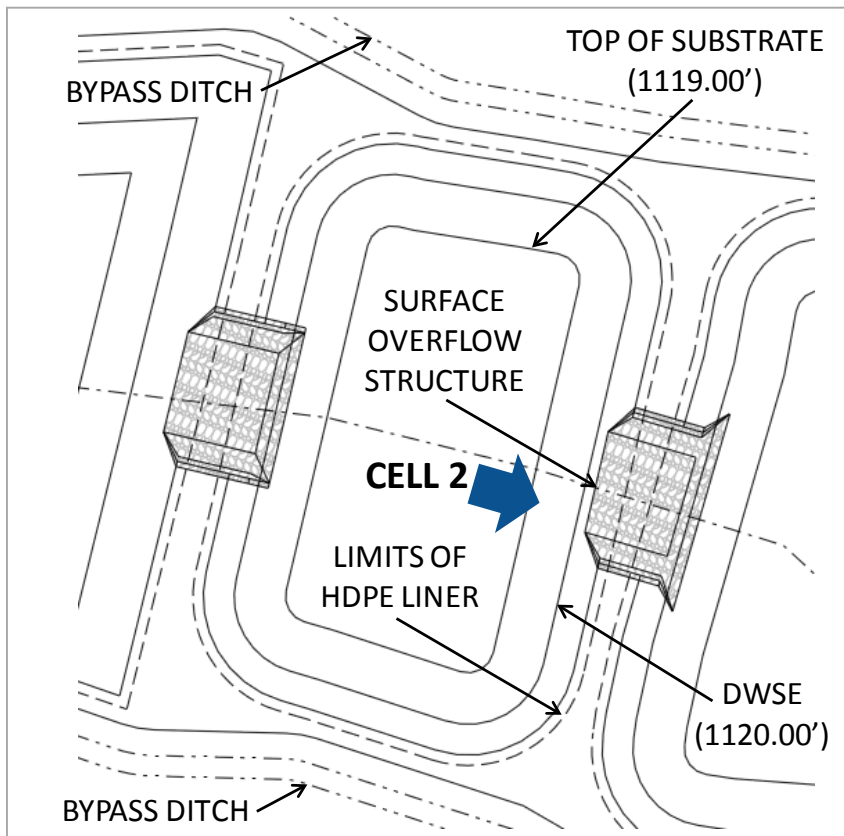
## Plan



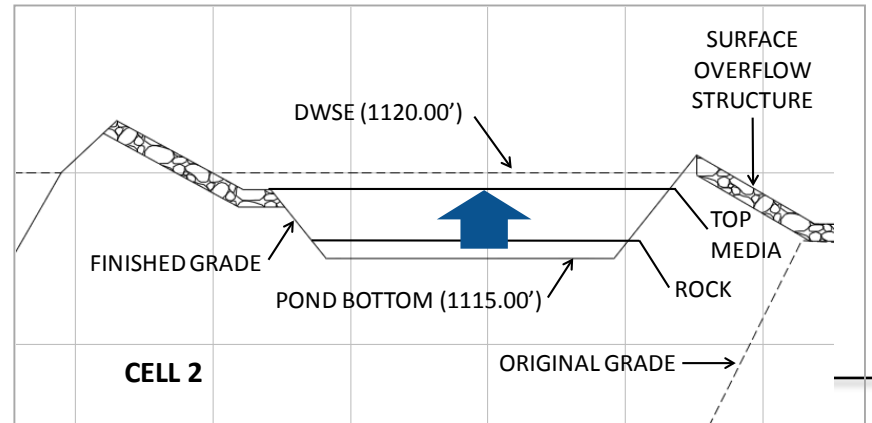
m <sup>2</sup>	Type	Media	Plants	Function
526	Downflow biochemical reactor	Mixed organic	None	Selenium reduction

# Cell 2: Upflow Anaerobic Wetland

## Plan



## Profile

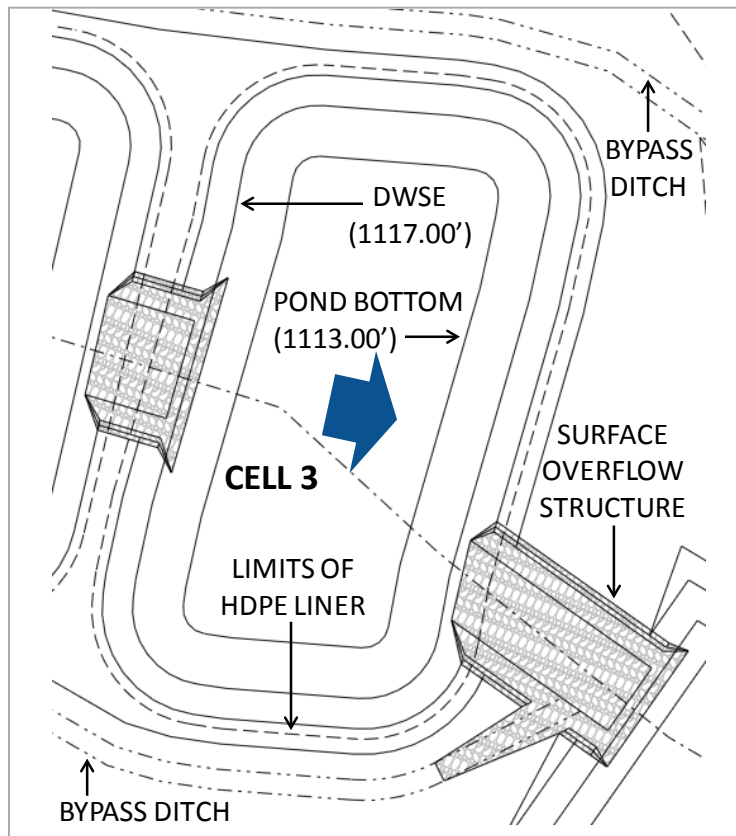


m <sup>2</sup>	Type	Media	Plants	Function
567	Upflow anaerobic	Peat	Sedges, rush	Selenium reduction, Byproduct polishing

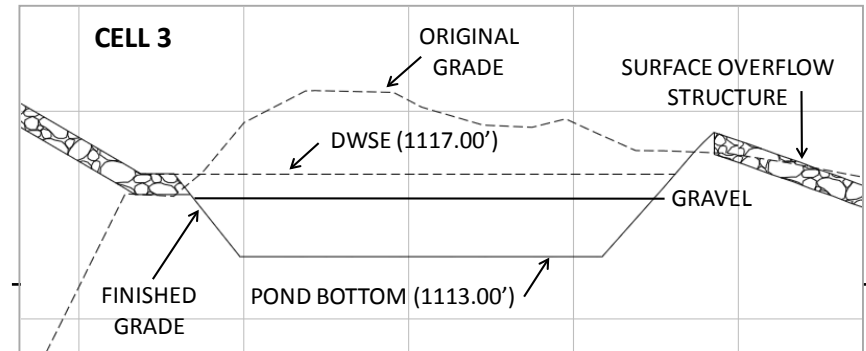


# Cell 3: Fill-and-Drain Polishing Wetland

## Plan



## Profile



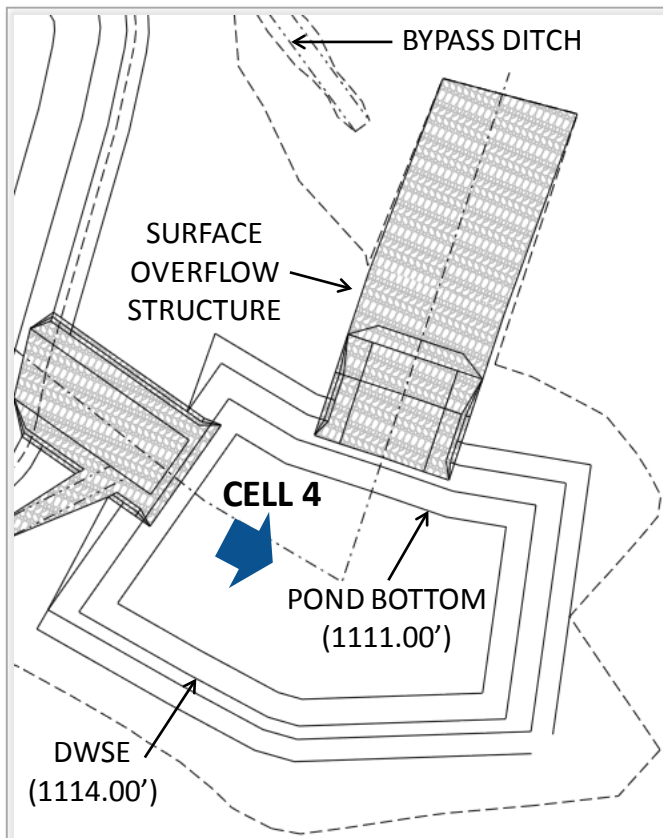
m <sup>2</sup>	Type	Media	Plants	Function
648	Subsurface fill and drain	Limestone gravel	Cattails	Byproduct polishing



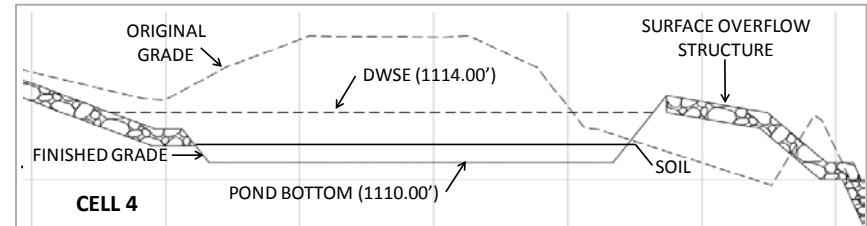


# Cell 4: Free Water Surface Polishing Wetland

## Plan



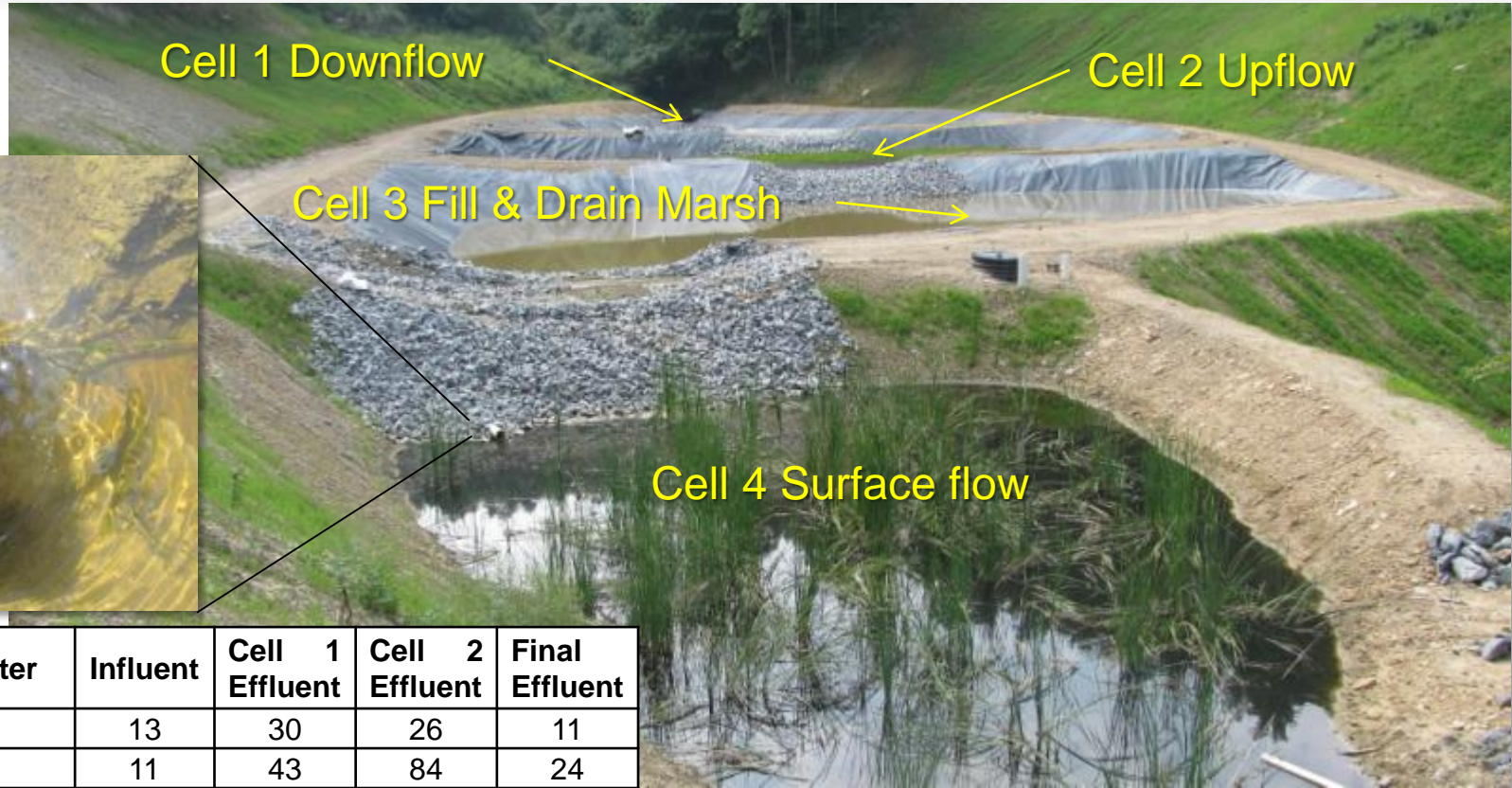
## Profile



m <sup>2</sup>	Type	Media	Plants	Function
445	Free water surface	Topsoil and ponded water	Cattails	Byproduct polishing



# PTS A: Completed Passive Se Treatment System



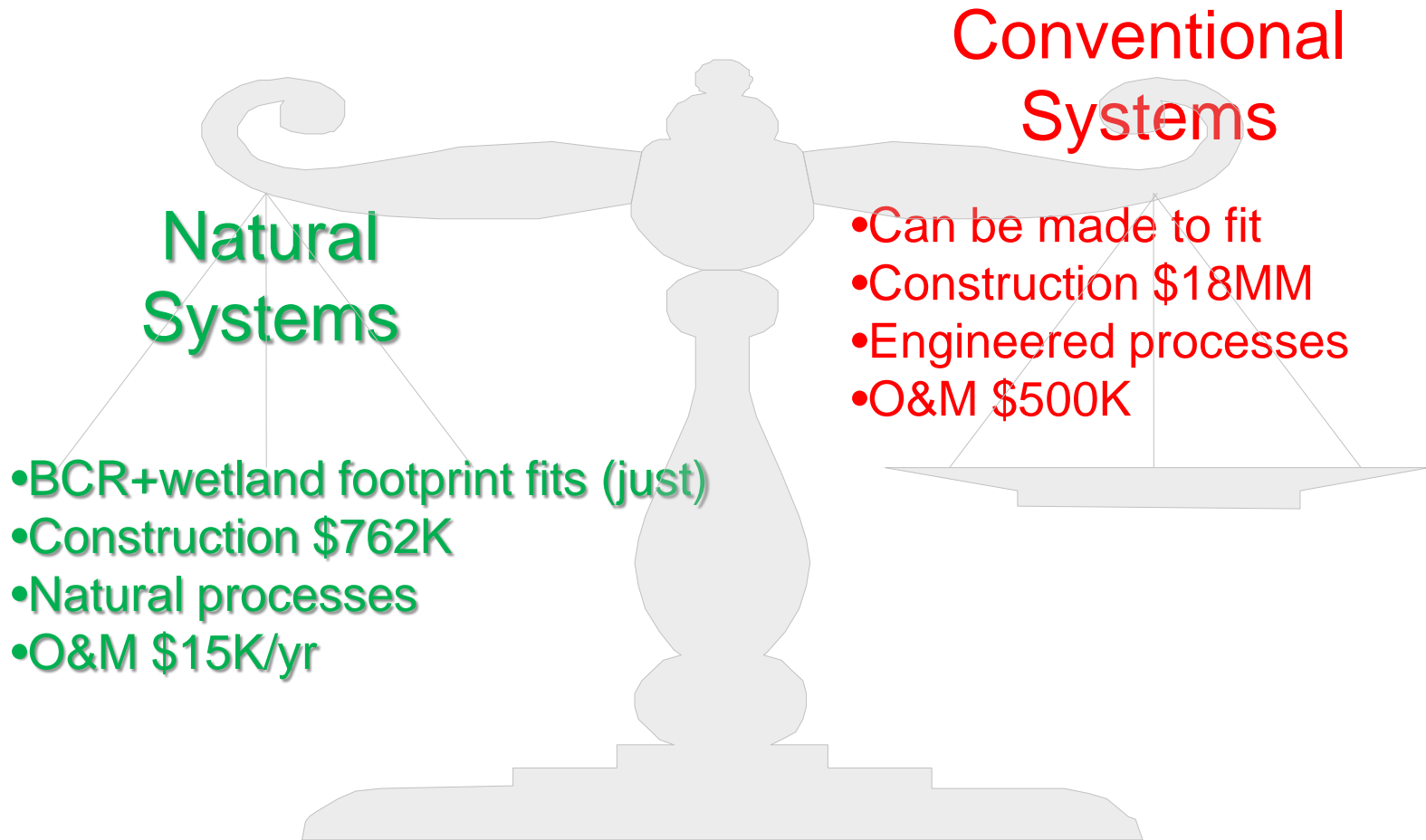
Parameter	Influent	Cell 1 Effluent	Cell 2 Effluent	Final Effluent
BOD	13	30	26	11
COD	11	43	84	24
NO <sub>2</sub> +NO <sub>3</sub> -N	3.6	1.5	2.4	1.2
Total Phosphorus	0.28	0.09	0.13	0.1

All units = mg/L

a. Monitoring data from February through July 2012

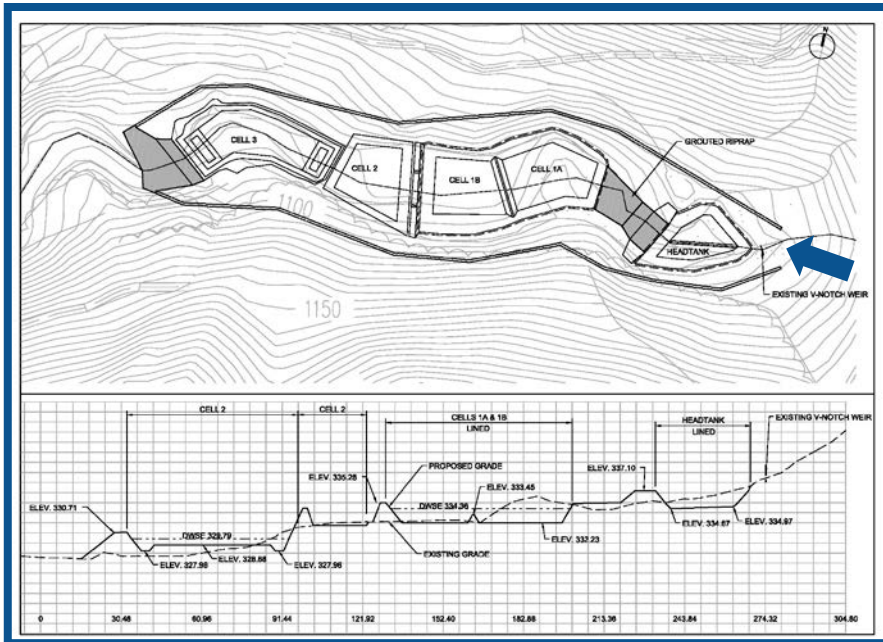
Source:  
Thomas, R. (2011)

# On Balance, Natural Systems Favored (Coal Mine Drainage Example)





# PTS B: Higher Flow, Higher Concentration



## ■ Five cells-in-series:

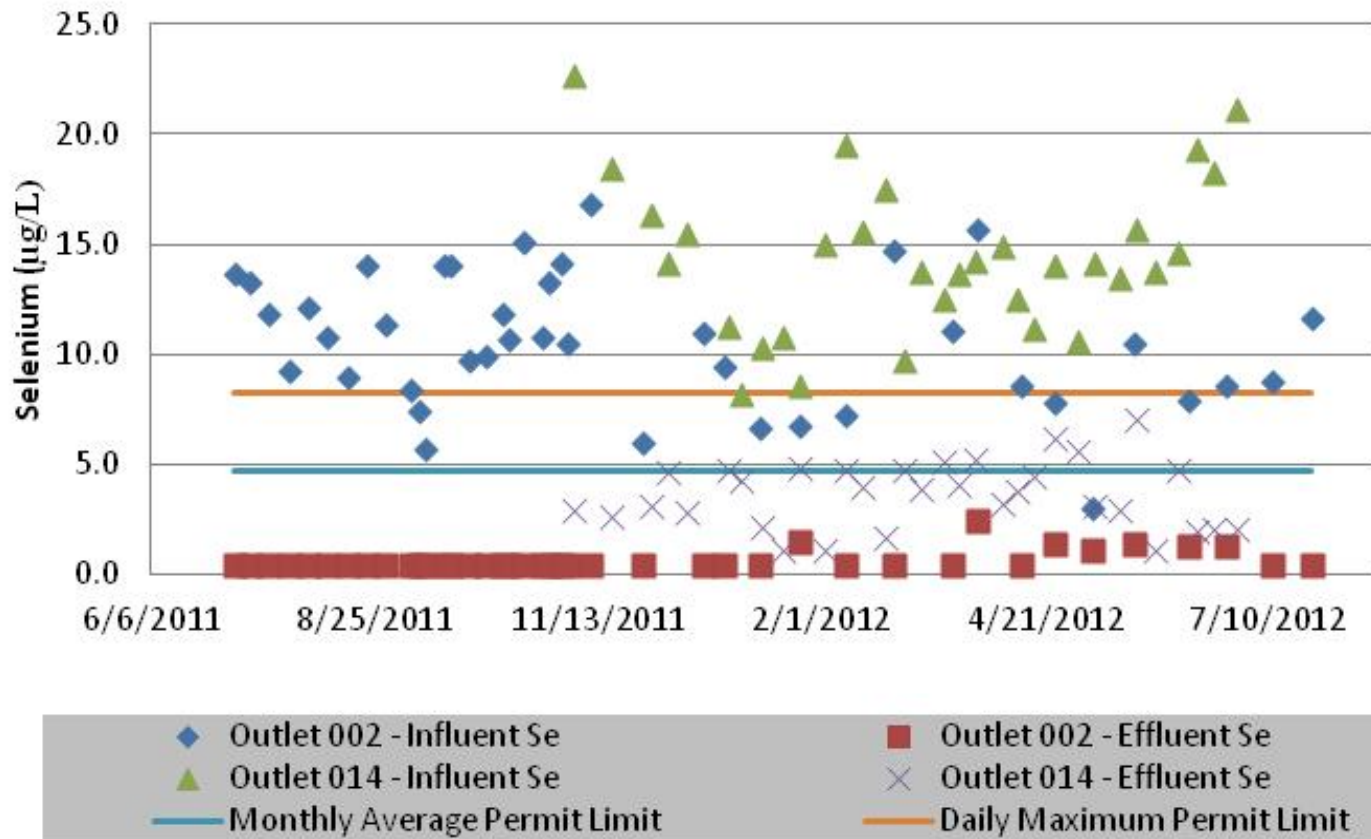
1. 0.12 ac Head tank
2. 0.48 ac Upflow BCR
3. 0.30 ac Upflow BCR
4. 0.23 ac Surface flow marsh
5. 0.38 ac Sedimentation pond

- 230 gpm base flow
- 24  $\mu\text{g}/\text{L}$  mean Se to  $<4.7$



# Selenium Treatment Performance Achieved WQ Targets

## Selenium Concentrations of Full-scale PTS Systems



Source:  
CH2MHILL (2012)

# Case History (2011-2012): Field-Scale Demonstrations for Coal Mine Drainage in WV



Source:  
J Bays (2011)

- Three reactors: 35 ft x50 ft
- Duration: 290, 203, 203 days
- Se in: 2-25  $\mu\text{g/L}$
- Se out: 1-4  $\mu\text{g/L}$
- Se RR: 0.22  $\text{mg/d/ft}^3$ 
  - » 7.7  $\text{mg/d/m}^3$
  - » 5-10°C
- Substrate: Haybales, MC
- Results used for full-scale plan:
  - 250,000  $\text{ft}^3$  substrate
  - 800 gpm
  - Se in: 14.88  $\mu\text{g/L}$
  - Se out: 2.35  $\mu\text{g/L}$
  - 10 hr HRT

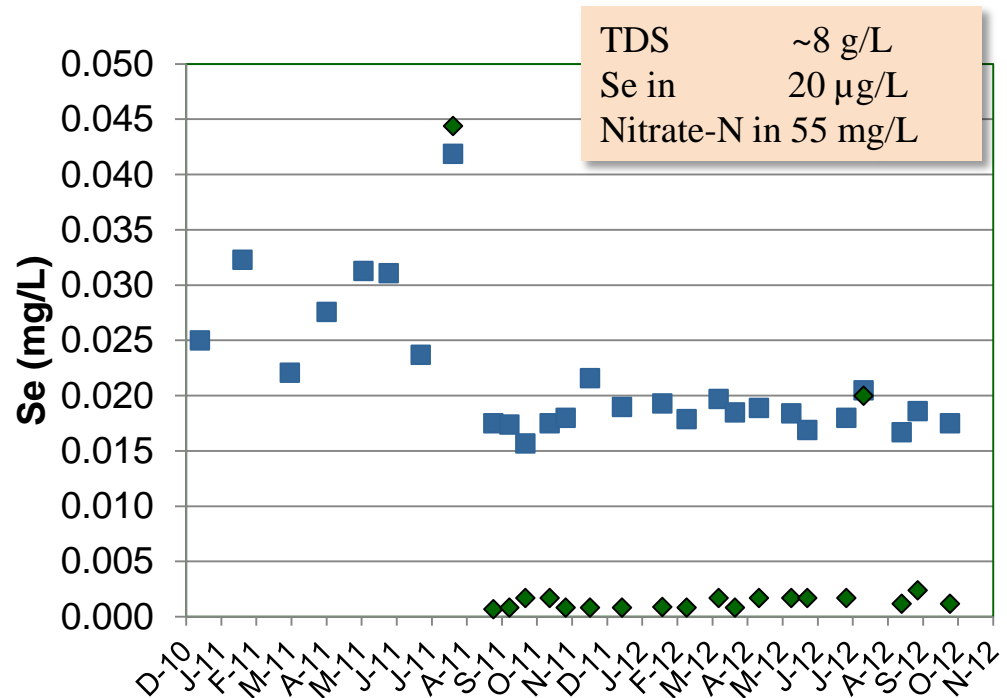
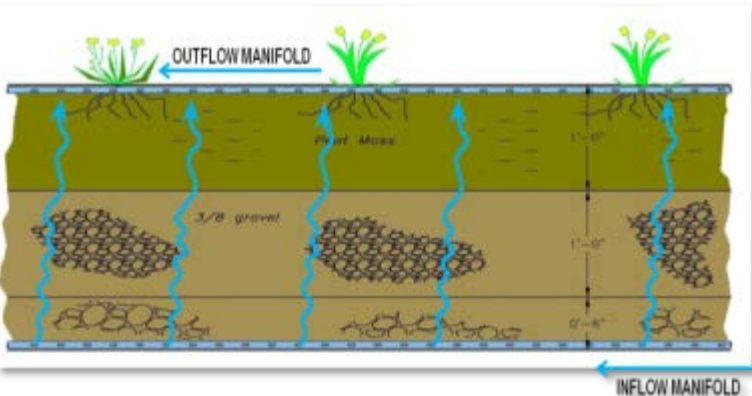
Source:  
Meek (2012)

# Case History (2008-present) Cold Climate Coal Mine, Alberta CA



- Date: 2008-present
- Type: Downflow VF
- Volume: 253 m<sup>3</sup>
- Temp in: 3.2°C avg
- Se in: 195 µg/L
- Se out: 33 µg/L (3 min)
- Se CR: >90%
- NO<sub>3</sub>N in: 36 mg/L
- NO<sub>3</sub>N RR: 5 g/d/m<sup>3</sup>
- Se RR: 17 mg/d/m<sup>3</sup>
- HRT: 4-8 days
- Year-round operation, passive

# Case History (2010-present) Continuous Se Removal in Mixed Organic Media for Saline RO Membrane Concentrate



■ Bin 2 Influent (RO Conc.)    ◆ Bin 2 Effluent

[www.usbr.gov](http://www.usbr.gov)



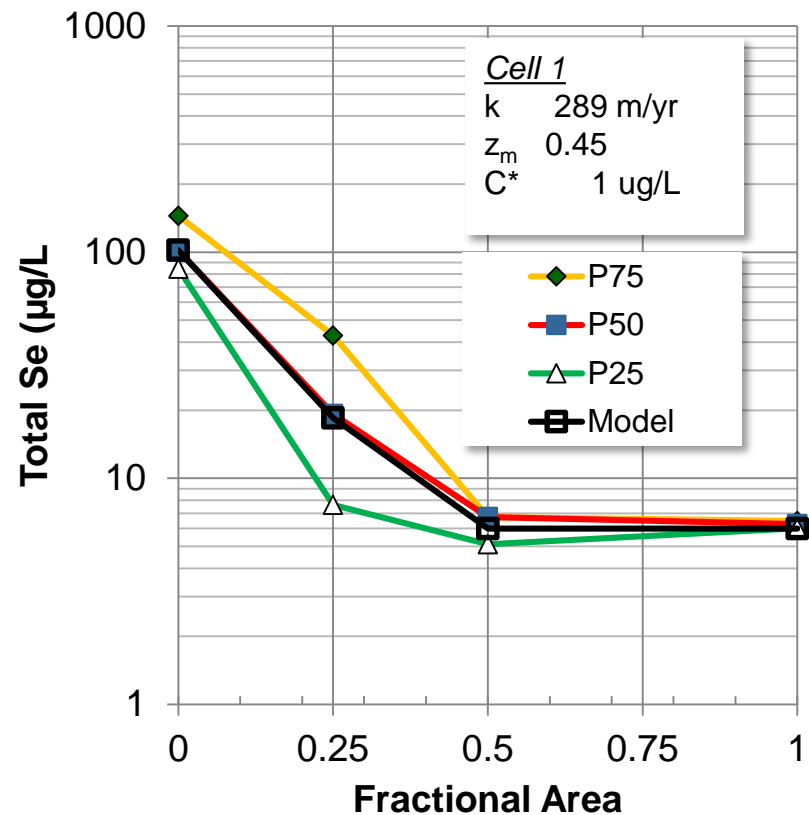
# Case History (2012): Treatment of Saline FGD Wastewaters Shows Selenium Removal

## Pilot Study Downflow Bioreactor



TDS	2-10 g/L
Se in	129 – 290 ug/L

## Treatment Trend: Se Profile

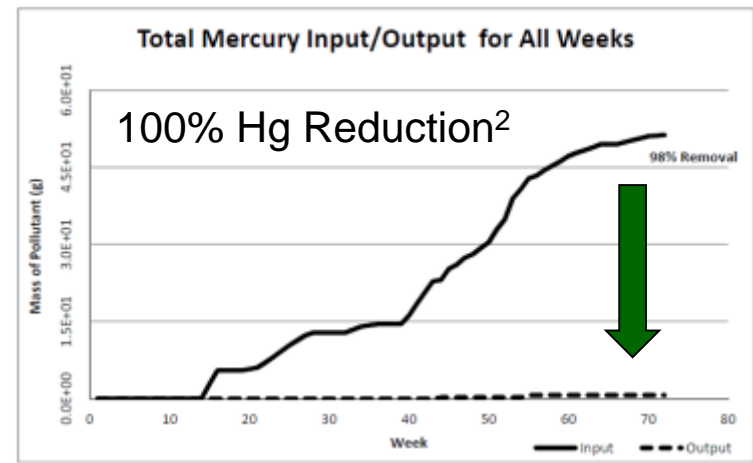
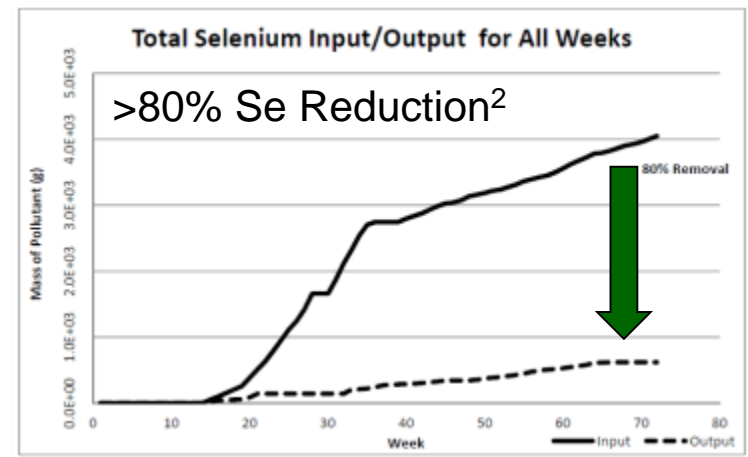


# Case History (2011-2012): Jeffrey Energy Center Pilot Wetland Westar Energy KS

## Vertical Flow Cells for Se, Hg Reduction<sup>1</sup>



TDS	~2 g/L
Se in	~70 µg/L
Area	2 ac



Sources:

<sup>1</sup> Morrison, J. (2012) [www.kdheks.com](http://www.kdheks.com)

<sup>2</sup> Talley, M. (2012).



# Passive Treatment of Selenium: BCR Byproducts

- BOD, COD, Low DO, Color, Nitrogen ( $\text{NH}_4^+$ ,  $\text{NO}_3^-/\text{NO}_2^-$ , TKN, etc), Phosphate, and Sulfide
- Recognition of the issue in early studies
- Why byproducts are an issue in Se treatment
- Expectations
  - Initial flush
  - Long-term generation of by products

# Functional Role of Aerobic Wetlands in Anaerobic + Aerobic Combination

## Surface Flow Wetlands



## Functions

- Treat BCR by-products
  - Oxidize BOD, COD
  - Trap particulates
  - Assimilate excess nutrients
  - Odor reduction
  - Reduce color
- Se polishing to trace levels
  - Biological vegetation uptake, transformation and burial
  - Hydrologic attenuation to equalize possible variation in flows and concentrations





# Aerobic Polishing Cells (APCs): How Well Do They Work?

## Conventional Parameters

Parameter	Removal Efficiency	Limit
BOD	50 – 90%	2 – 10 mg/L
TSS	50 – 90%	2 – 10 mg/L
TN	50 – 90%	1 – 3 mg/L
TP	40 – 90%	< 1 mg/L

Note: Removal efficiencies and effluent concentrations depend on influent concentration and hydraulic loading rate.



# Conclusions

- Se control can be critical to agriculture runoff, power generation, and mining industries
- Se control includes active and passive technologies
- Passive treatment:
  - Uses natural processes to reduce and capture Se
  - Requires less operational effort and management
  - Demonstrated as a viable option for Se treatment
- Site-specific applications based on Se concentration, flow rate, topography, and general influent geochemistry
- Early success achieved; optimization ongoing as systems age and performance is evaluated through changing climatic and flow conditions



# Acknowledgements

- Thanks to all of our collaborating partners in the West Virginia Coal Mining Industry
- Thanks to supporting engineering and science staff at CH2M HILL



# Questions